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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/591,233	05/07/2007	Malin Ernebrant	02508.0110	9327
FINNEGAN, HENDERSON, FARABOW, GARRETT & DUNNER LLP 901 NEW YORK AVENUE, NW WASHINGTON, DC 20001-4413			EXAMINER	
			FISHER, ABIGAIL L	
			ART UNIT	PAPER NUMBER
			1616	
			MAIL DATE	DELIVERY MODE
			06/13/2011	PAPER

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

Advisory Action Before the Filing of an Appeal Brief

Application No.	Applicant(s)
10/591,233	ERNEBRANT ET AL.
Examiner	Art Unit
ABIGAIL FISHER	1616

	ABIGAIL FISHER	1616				
The MAILING DATE of this communication appea	rs on the cover sheet with the o	correspondence address				
THE REPLY FILED <u>18 May 2011</u> FAILS TO PLACE THIS APPL	ICATION IN CONDITION FOR AL	LOWANCE.				
1. The reply was filed after a final rejection, but prior to or on tapplication, applicant must timely file one of the following rapplication in condition for allowance; (2) a Notice of Appefor Continued Examination (RCE) in compliance with 37 Caperiods:	eplies: (1) an amendment, affidavi al (with appeal fee) in compliance	t, or other evidence, which places the with 37 CFR 41.31; or (3) a Request				
a) \boxtimes The period for reply expires <u>3</u> months from the mailing date of the final rejection.						
b) The period for reply expires on: (1) the mailing date of this Ac no event, however, will the statutory period for reply expire la Examiner Note: If box 1 is checked, check either box (a) or (b	er than SIX MONTHS from the mailin). ONLY CHECK BOX (b) WHEN THE	g date of the final rejection.				
MONTHS OF THE FINAL REJECTION. See MPEP 706.07(f) Extensions of time may be obtained under 37 CFR 1.136(a). The date of have been filed is the date for purposes of determining the period of exteunder 37 CFR 1.17(a) is calculated from: (1) the expiration date of the shat forth in (b) above, if checked. Any reply received by the Office later that may reduce any earned patent term adjustment. See 37 CFR 1.704(b). NOTICE OF APPEAL	n which the petition under 37 CFR 1.1 nsion and the corresponding amount ortened statutory period for reply origi	of the fee. The appropriate extension fee nally set in the final Office action; or (2) as				
2. The Notice of Appeal was filed on A brief in compli	ance with 37 CFR 41.37 must be	filed within two months of the date of				
filing the Notice of Appeal (37 CFR 41.37(a)), or any exten a Notice of Appeal has been filed, any reply must be filed v AMENDMENTS	sion thereof (37 CFR 41.37(e)), to	avoid dismissal of the appeal. Since				
3. The proposed amendment(s) filed after a final rejection, b	ut prior to the date of filing a brief,	will not be entered because				
(a) They raise new issues that would require further con	•	TE below);				
(b) They raise the issue of new matter (see NOTE below	•					
(c) They are not deemed to place the application in bette appeal; and/or	er form for appeal by materially re	ducing or simplifying the issues for				
(d) They present additional claims without canceling a co	orresponding number of finally rej	ected claims.				
NOTE: (See 37 CFR 1.116 and 41.33(a)).						
4. The amendments are not in compliance with 37 CFR 1.12	1. See attached Notice of Non-Co	mpliant Amendment (PTOL-324).				
5. Applicant's reply has overcome the following rejection(s):	<u> </u>					
	S. Newly proposed or amended claim(s) would be allowable if submitted in a separate, timely filed amendment canceling the					
non-allowable claim(s). 7. For purposes of appeal, the proposed amendment(s): a) how the new or amended claims would be rejected is provious The status of the claim(s) is (or will be) as follows: Claim(s) allowed: Claim(s) objected to: Claim(s) withdrawn from consideration:		l be entered and an explanation of				
AFFIDAVIT OR OTHER EVIDENCE						
8. The affidavit or other evidence filed after a final action, but before or on the date of filing a Notice of Appeal will <u>not</u> be entered because applicant failed to provide a showing of good and sufficient reasons why the affidavit or other evidence is necessary and was not earlier presented. See 37 CFR 1.116(e).						
9. The affidavit or other evidence filed after the date of filing a Notice of Appeal, but prior to the date of filing a brief, will <u>not</u> be entered because the affidavit or other evidence failed to overcome <u>all</u> rejections under appeal and/or appellant fails to provide a showing a good and sufficient reasons why it is necessary and was not earlier presented. See 37 CFR 41.33(d)(1).						
10. 🗌 The affidavit or other evidence is entered. An explanation of the status of the claims after entry is below or attached.						
REQUEST FOR RECONSIDERATION/OTHER 11. ☑ The request for reconsideration has been considered but does NOT place the application in condition for allowance because:						
See Continuation Sheet.						
12. ☑ Note the attached Information <i>Disclosure Statement</i> (s). (PTO/SB/08) Paper No(s). <u>1/21/11</u> 13. ☐ Other:						
/Johann R. Richter/	6/9/11					
Supervisory Patent Examiner, Art Unit 1616						

Continuation of 11. does NOT place the application in condition for allowance because: The rejections are maintained for the reasons set forth in the Final Office action. Applicants argue that it the examiner has not shown why it would have been obvious when Zander and Duponchelle are considered as a whole to manipulate he pH of the alkaline solution to the claimed pH of 10.1 to 10.5. The examiner disagrees. The reasons for this manipulation are clearly set in in pages 6-7 of the final Office action. Specifically, based on the teachings of Zander and Duponchelle et al. it would have been obvious to one of ordinary skill in the art to manipulate the pH in order to produce a more stable bicarbonate solution that does not need the use of special equipment to keep the carbon dioxide concentration constant while maintaining the partial pressure of the carbon dioxide to that is of the same magnitude as the atmosphere. Applicants argue that it is the particular proportion of bicarbonate/carbonate in the first single solution that allows it to be in equilibrium with the partial pressure of carbon dioxide in the atmosphere and Zander fails to recognize that it is the specific proportions of carbonate/bicarbonate that causes this equilibrium. This argument is not persuasive, while the claims do recite the equilibrium of partial pressure, the claims do not require any specific proportions. Zander recognizes the desire to have the magnitude of the carbon dioxide of the solution match that of the atmosphere. Therefore, manipulation to achieve this matching would have been obvious. Applicants argue that the ratio of bicarbonate to carbonate taught by Zander would not be expected to be in equilibrium with the partial pressure of atmospheric carbon dioxide. However, applicants have not shown this to be true. "The arguments of counsel cannot take the place of evidence in the record." In re Schulze, 346 F.2d 600, 145 USPQ 716, 718 (CCPA 1965), In re Huang, 40 USPQ 2d 1685 (Fed. Cir. 1996), In re De Blauwe et al., 222 USPQ 191, (Fed. Cir. 1984). Applicant has not provided any factual evidence establishing unobviousness. Since Zander teaches this matching of the carbon dioxide pressure, manipulation is still obvious. Applicants argue that the examiner's contention that Zander would be expected to have the same stability as instantly claimed is baseless. The examiner disagrees. Firstly it is noted that neither stability nor time before mixing of the two solutions is claimed. Secondly, the reason the examiner has stated that there is a reasonable expectation the stability would be the same or similar is because applicants have stated this stability is due to the partial pressure of carbon dioxide which is taught by Zander. Applicants argue that the stability due to the partial pressure could not have been predicted. The examiner disagrees. Zander teaches that instability of these types of solutions is due to the escape of carbon dioxide. Zander teaches liquids which have a carbon dioxide partial pressure which corresponds to that of atmospheric air do not change their overall carbon dioxide content which makes them more stable. Therefore, demonstration of this statement does not make the results unpredictable. It is for at least these reasons the rejections are maintained for the reasons set forth in the Final Office action.